

 <b>MLF Experimental Report</b>		提出日 Date of Report July 24, 2012
課題番号 Project No.	2012A0089	装置責任者 Name of responsible person Yasuhiro Miyake
実験課題名 Title of experiment	Static/dynamic behavior in MgH <sub>2</sub> at high temperatures	装置名 Name of Instrument/(BL No.) D1
実験責任者名 Name of principal investigator	Jun Sugiyama	実施日 Date of Experiment June 06, 2012 – June 10, 2012
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試料、実験方法、利用の結果得られた主なデータ、考察、結論等を、記述して下さい。(適宜、図表添付のこと)

Please report your samples, experimental method and results, discussion and conclusions. Please add figures and tables for better explanation.

1. 試料 Name of sample(s) and chemical formula, or compositions including physical form.
Two powder samples (MgH <sub>2</sub> and milled MgH <sub>2</sub> ), which were prepared at Toyota CRDL, were packed into a gold O-ring sealed titanium cell in an Ar-filled glove box. Then, the Ti cell was mounted on a graphite sample holder in order to measure wTF-, ZF- and wLF- $\mu$ SR spectrum at temperatures from 300 to 650 K under high vacuum. In order to avoid an unexpected explosion at high T due to the desorption reaction, we made small holes on the window— i.e. a Ti foil with 50 $\mu$ m thickness, of the Ti cell just before the measurements.

2. 実験方法及び結果 (実験がうまくいかなかった場合、その理由を記述してください。)
Experimental method and results. If you failed to conduct experiment as planned, please describe reasons.
<p>Following upon the <math>\mu</math>SR measurements on a hydrogen storage material, MgH<sub>2</sub> below 500 K, we have measured wTF-, ZF-, and wLF-<math>\mu</math>SR spectra for an “as prepared MgH<sub>2</sub>” and “milled MgH<sub>2</sub>” sample until 750 K, in order to study the change in <math>\mu</math>SR parameters by the H-desorption reaction. Particularly, although there are no crucial changes in crystal structure between the “as prepared MgH<sub>2</sub>” sample and “milled MgH<sub>2</sub>” sample, the hydrogen desorption temperature (<math>T_d</math>) for the latter sample decreases by about 50 K compared with <math>T_d</math> for the former sample [1].</p> <p>Since the hydrogen desorption reaction of MgH<sub>2</sub> is an endothermal reaction, it was very difficult to control the sample temperature around <math>T_d</math>. In particular, the infrared lamp used as a heater of the present oven was found to respond too fast to stabilize the temperature around <math>T_d</math>. Therefore, for the milled sample, we obtained reliable data only below 570 K, whereas <math>T_d = 620</math> K. Figures 1(a) and 1(b) show the temperature dependences of the relaxation rate (<math>\lambda_{TF}</math>) of the wTF-spectrum, which corresponds to a spin-spin relaxation rate, i.e., the field distribution width (<math>\Delta</math>) due to <sup>1</sup>H nuclear field, and the field fluctuation rate (<math>\nu</math>) for the two samples. For the as prepared sample, both <math>\lambda_{TF}</math> and <math>\nu</math> are almost temperature independent up to 630 K, and then become very small, eventually 0 above <math>T_d</math>. This means</p>

## 2. 実験方法及び結果(つづき) Experimental method and results (continued)

that the H-desorption reaction abruptly occurs and completes at  $T_d$ . In fact, at temperatures above  $T_d$ , the wTF-spectrum exhibited a non-relaxing oscillatory signal due to the applied wTF. Furthermore, the wTF-spectrum obtained in cooling mode from 750 K did not change even below  $T_d$ , suggesting that all the hydrogen atoms were removed from the sample.

On the other hand, for the milled sample,  $\lambda_{TF}$  gradually decreases with temperature above 470 K, while  $\nu$  apparently increases with temperature also above 470 K. This suggests that the H-desorption reaction starts to occur far below  $T_d$  (=620 K) for the milled sample. Moreover, the increase in  $\nu$  implies rapid diffusion of  $H^+$  and/or  $\mu^+$  in the sample, which is proposed to be a significant factor to decrease  $T_d$  for the other hydrogen storage materials, such as  $LiAlH_4$  [2] and  $M(BH_4)_n$  [2].

However, in order to further elucidate the mechanism on the decrease in  $T_d$  by milling and/or by adding a small amount of  $Nb_2O_5$ , we need more systematic measurements at high temperatures, particularly below the vicinity of  $T_d$  using a more reliable temperature control system.

### REFERENCES

- [1] N. Hanada, et al., J. Alloys and Compounds **446-447**, 67 (2007).
- [2] R. Kadono, et al., Phys. Rev. Lett. **100**, 026401 (2008).
- [3] J. Sugiyama, et al., Phys. Rev. B **81**, 092103 (2010).

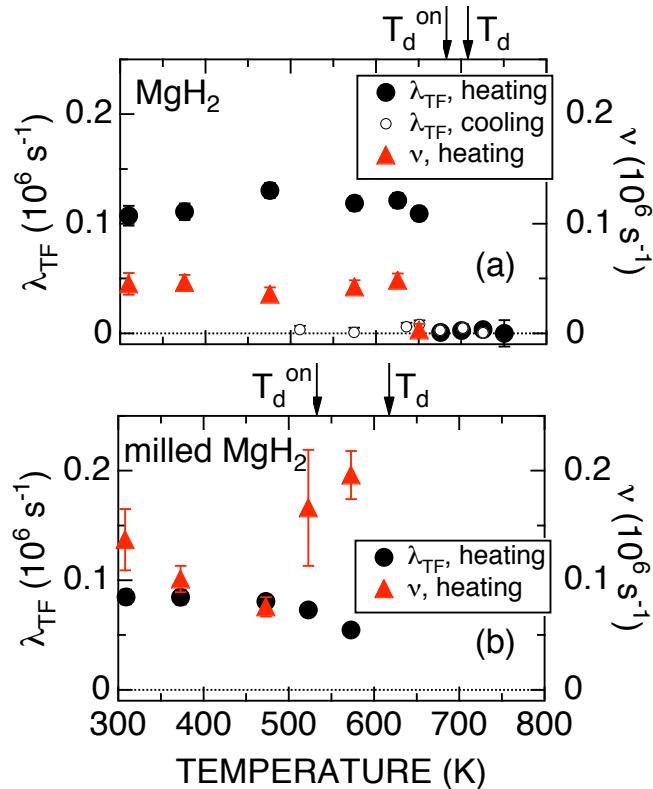


Fig. 1: The temperature dependences of the wTF relaxation rate ( $\lambda_{TF}$ ) and the field fluctuation rate ( $\nu$ ) for (a) the as prepared  $MgH_2$  and (b) milled  $MgH_2$  sample. The data of  $\lambda_{TF}$  were obtained by fitting the wTF-spectrum using an exponentially relaxing cosine oscillation function, and that of  $\nu$  were obtained by fitting the ZF- and wLF-spectra using a dynamic Kubo-Toyabe function. Here, wTF was 20 Oe and wLF was 10 Oe.